

A PAIR OF ATOMS WITH THE SAME STRUCTURE BUT WITH OPPOSITELY ROTATING PROTONS

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ABSTRACT: Using the ring model of atoms and molecules, was revealed that the remust be two type A and type B) of atoms that differ in the direction of rotation of individual rings (for type B, all rings rotate opposite to the directions of rotation for type A) but have the same structure. The rotations of protons and neutrons in atoms are given by a system that resembles a "cogwheel" in gear box. Both types A and B have the same structure. Solution to this problem was found in the design of the TiO2 model, which has a total of 22 protons, 22 electrons and 26 neutrons in titanium atom (Ti). This also applies to the oxygen molecule. Both oxygen atoms have the same structure, but the individual protons rotate in opposite directions in both atoms. Magnetic lines of force with opposite orientations are then created in the proton axis. Only such oxygen atoms can combine to form an oxygen molecule. Magnetic lines of force emerge from one proton of the first oxygen atom and can thus enter the proton of the second oxygen atom. Each oxygen atom has 8 protons, 8 electrons and 8 neutrons.

KEYWORDS: Ring model, Directionofrotationof proton, Neutron and electron, Directions of magnetic moments, photocatalytic cleaner with TiO₂, Graphene, Photocatalysis, Titanium dioxide TiO₂, UV light.

INTRODUCTION

Currently, children in our schools are taught that the nucleus of atoms consists of balls of protons and neutrons. Electrons orbit around such a nucleus. This model was inspired by the planetary system, where individual planets (including the Earth) orbit the sun[1,5]. This Bohr model cannot explain, for example, the hydrogen molecule, which consists of two hydrogen atoms. This model needs to be modified. Electrons, protons are not balls, but rings with a fractal structure, but in the hydrogen atom, the electron is attracted by an electric force and repelled by a magnetic force. In the Bohr atom, the repulsive force is created by a centrifugal force. What is interesting about both models of hydrogen atoms is that the distances between the proton and the electron are the same in both models.

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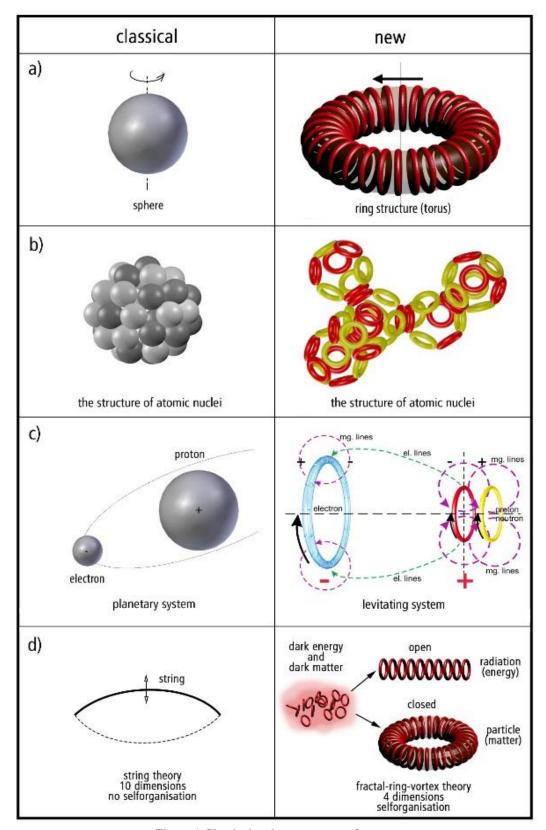


Figure 1 Classical and newstructureofatoms

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The classical approach in particle physics is based on the fact that the electron has some parameters like charge, mass, etc. but does structure. Theelectroniscalculated as point particlehavingmagneticproperties. Combiningknowledgeofphysicalchemistry [1-3]. Evolutionaryoptimization, 3Dgraphics, programming in Python, and to create1 programsfordesigningnewnanostructuremodels. Thefirst testing mathematics make itpossible proposal for the nanostructure prediction program is limited to carbon structures. On Figure 1 are classical and newmodelsofNature. Theory VFRT (VortexFractal Ring Theory) usestheelectron, proton, and neutron as theparticlewith a toroidal (ring) shape, whichisformed by fractalsubstructures [4]. connected to eachother by vortexelectromagnetic fields. Theatomicnucleuscanbebuiltfrom ring protons and neutrons. Combiningknowledgeofatomicnucleusstructures and evolutionaryoptimizationmakesitpossible to createprogramsfordesigningnewnanostructuremodels. proposal for the nanostructure prediction program was limited to carbon structures. The aimwas to verify whether the proposed program iscapableofgeneratingknowncarbon na nostructures, such as graphene. Thefollowingversionsofthe program will no longerhavethislimitation.

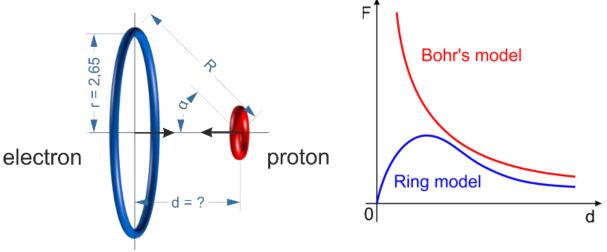


Figure 2Levitating ring electron in the hydrogen atom



Figure 3 Basic substructures of the atomic nucleus.

Theleftsubstructureis a helium nucleus (alphaparticle). Therightsubstructureis notused in theperiodic table of elements. Thethreesubstructures in themiddle are key.

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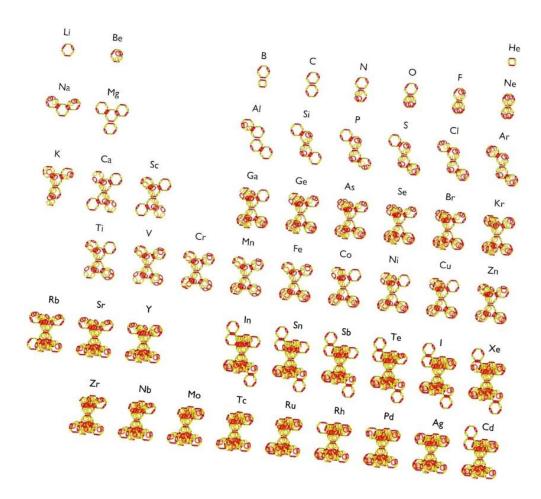


Figure4Theperiodic table of elements.

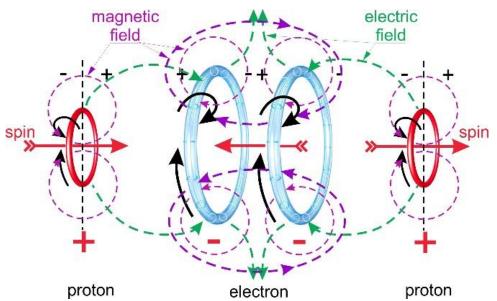


Figure 5 Hydrogen moleculewithcovalent bond (topologicalstructure – rings are not in rightscale).

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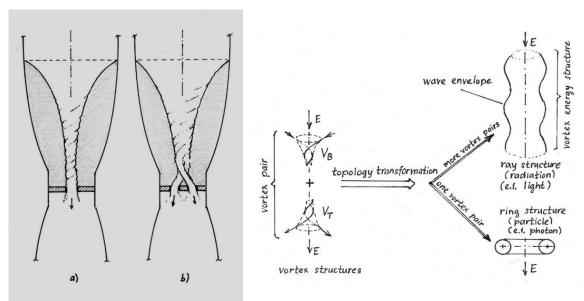


Figure6Vortexstructures(structure of light as a ring particle or a wave energy structure)

MATHEMATICAL DESCRIPTION

Electronradius r_e and spin S. m_e isthemassoftheelectron, v_e isthecircumferentialvelocityoftheelectron ring, and lambda isthewavelength(de Broglie'sequation)

$$r_e = \frac{\mu_o e^2}{8\pi^2 m_e} \bullet \frac{1}{v_e^2} \vec{S} = m_e (\vec{r}_e \times \vec{v}_e)$$

$$|S_z| = N \frac{m_e}{N} r_e v_e = \frac{1}{2} \frac{h}{2\pi} 2 \cdot 2\pi r_e = n\lambda$$

$$\lambda = \frac{h}{m_e v_e}$$



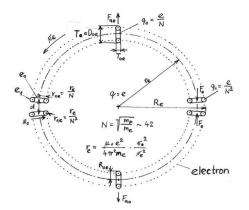


Figure7 Electron

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$$\begin{split} \vec{S} &= m_{e}(\vec{r}_{e} \times \vec{v}_{e}) \\ S_{z} &= N \frac{m_{e}}{N} r_{e} v_{e} \\ r_{e} &= \frac{e^{2}}{8\pi^{2} \varepsilon_{0} m_{e}} \cdot \frac{1}{v_{e}^{2}} \\ v_{e} &= \frac{e^{2}}{2\pi \varepsilon_{0} h} \\ r_{e} &= \frac{e^{2}}{8\pi^{2} \varepsilon_{0} m_{e}} \cdot \frac{1}{v_{e}^{2}} = \frac{e^{2}}{8\pi^{2} \varepsilon_{0} m_{e}} \cdot \frac{4\pi^{2} \varepsilon_{0}^{2} h^{2}}{e^{4}} = \frac{\varepsilon_{0} h^{2}}{2m_{e} e^{2}} \\ S_{z} &= \pm m_{e} v_{e} r_{e} = \pm m_{e} \frac{e^{2}}{2\pi \varepsilon_{0} h} \cdot \frac{\varepsilon_{0} h^{2}}{2m_{e} e^{2}} = \pm \frac{1}{2} \cdot \frac{h}{2\pi} = \pm \frac{1}{2} \hbar = m_{s} \hbar \\ m_{s} &= \pm \frac{1}{2} \end{split}$$

$$\begin{split} M_z &= IS \\ T &= \frac{2\pi r_e}{\overline{v}_e} \\ I &= \pm \frac{Q}{T} = \pm \frac{\frac{e}{N}N}{\frac{2\pi r_e}{\overline{v}_e} \cdot \frac{1}{2}} = \pm \frac{e\overline{v}_e}{\pi r_e} \\ S &= \pi r_e^2 \\ M_z &= IS = \pm \frac{e\overline{v}_e}{\pi r_e} \cdot \pi r_e^2 \cdot \frac{m_e}{m_e} = \pm \frac{e}{m_e} S_z = \pm \frac{e}{m_e} \cdot \frac{1}{2}\hbar = \pm \mu_B \end{split}$$

The spin Szof the electron and magnetic momentum Mz

A pair of atoms with the same structure

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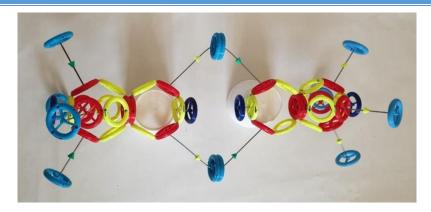
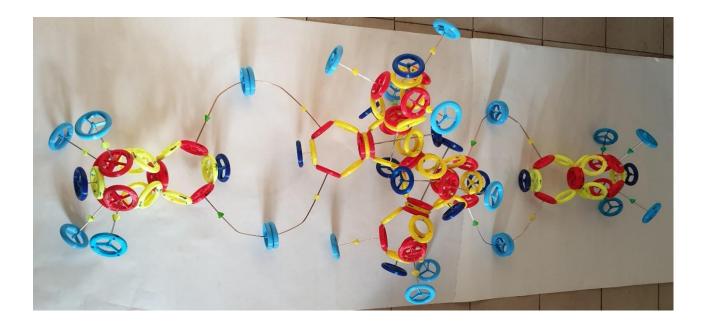


Figure 8 Oxygenmoleculewithoppositelyrotatingprotons(type A and type B). Both oxygen atomsin Figure 8 have the same structure. The direction of rotation of the rings differs. For bonding, it is important to have both types, because a magnetic field leaves one atom and enters the other.

Figure 7 The TiO2 model with oppositely rotating protons (type A and type B).



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Therotationsofprotons and neutrons in atoms are given by a systemthatresembles a "cogwheel" in gear box. Bothtypes A and B have the same structure. The titanium atom Ti model has a total of 22 protons, 22 electrons and 26 neutrons. Each oxygen atom has 8 protons, 8 electrons and 8 neutrons. The TiO2 model thereforeconsistsof 118 rings (protons, neutrons and electrons). Twotypesofelectrons are distinguished in the model (valence electrons are coloredlight blue and innerelectrons are coloreddark blue). The direction of rotation of the proton determines the direction of the magnetic moment that exits the atom. Foranexternalviewofthe proton, it has (forclockwiserotationofthe proton) the proton has a magnetic moment that exits the proton (yellowarrow). Forcounterclockwiserotationofthe proton) the proton has a magnetic moment thatentersthe proton (green arrow). For example, in the case of an oxygen molecule O2, oxygen atoms of type A and type B are joined. Themagneticmoments in the binding of two protons must follow the same direction.Onlythemagnetic withtheyellowarrowcanbecombinedwiththemagnetic withthe (Figure 9). Theair moment green arrow cleanerusesfamiliarprinciples but in anunconventionaloriginalway. Thesurfacewithtitanium dioxide TiO2 isactivated by UV light (a wavelengthof 365 nm - band A withthe maximum photocatalysisefficiency). The air flowis done by 2 fansthathaveanadjustable speed. Photocatalysisbreaksdownviruses (including coronavirus and allitscurrent and futuremutations) intoharmlessatoms and molecules. Theactivatedsurfaceoftitanium dioxide (TiO2) lightwith a wavelengthofabout 365nm (band A) isused to removesbindingelectronsfromthestructures. UV activatethephotocatalyticsurface, in contrast to hard radiationwith a wavelengthofabout 200nm (band C). Hard radiationkillslivingstructures but does not breakdownintosimplerstructures. We use UV lightgenerated by UV. Theadvantageof UV LEDsisthat a low and safe DC supplyvoltageof 12 or24Vcanbeused. Wecanonly hope thatthisprinciplewillhelpslowdownor stop the spread of viral diseases. Photocatalysis also decomposes harmful bacteria, fungi, unpleasantodors, cigarettesmoke and chimneysmoke, and harmfulgasesfromcars and motorcycles.

CONCLUSION

All atomsexist in twotypes (type A and B). They havethesamestructure but haveoppositemagneticmoments. All models are topologicalstructures. The fractal model of the proton, neutron and electron allows us to explain what dark matter and energy could be. Small fractal substructures can be part of dark matter and energy. If they form closed structures (rings) they are matter. If they form open chains, they are energy. Unfortunately, these small fractal substructures cannot be observed with light. Dark matter (see Figure 1d new) manifests itself, for example, in the equal rotation rate of the inner and outer parts of galaxies.

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